## Virginia Division of Consolidated Laboratory Services

SULFATE ION IN WATER by ASTM D516-02						
Facility Name:	VELAP ID					
Assessor Name: Analyst Name:		Ins	spect	tion Da	ite	
Relevant Aspect of Standards	Method Reference	Y	N	N/A	Comments	
Records Examined: SOP Number/ Revision/ Date				Ar	nalyst:	
Sample ID: Date of Sample Prepa	ration:		Dat	te of A	nalysis:	
If a spectrophotometer was used, was it used at 420 nm with a light path of 4 to 5 cm?	7.1.2					
If a filter photometer was used, was it used with a violet filter having a maximum near 420 nm and a light path of 4 to 5 cm?	7.1.3					
Were samples filtered if turbid?	11.1					
Were sample temperatures adjusted to between 15 and 30°C prior to analysis?	11.1					
Were Barium Chloride crystals of between 20 and 30 mesh?	8.3					
After the addition of Barium Chloride crystals to the samples, were they stirred at a constant speed for exactly 1 minute?	11.5					
Immediately after stirring period ended, were sample turbidities measured at 30 second intervals for 4 minutes, and the maximum turbidity during that period recorded?	11.6					
If interferences were suspected in a sample, were interferences ruled out by analyzing a two-fold dilution of a sample to have one half of the sulfate determined in the undiluted sample?	11.8					
Was a calibration standards analyzed by the same process as the samples?	10.1					
Were sample sulfate concentrations calculated against a calibration curve?	12.1					
Was the calibration curve composed of a calibration blank and at least four calibration standards?	14.1.1					
Notes/Comments:						

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Did the calibration curves have a correlation coefficient ≥0.990?	14.1.1				
Was the <b>calibration blank</b> analyzed at the end of each batch run?	14.1.1				
Was an instrument check standard analyzed at a rate of 10% of samples and to fall between 80 and 120% of expected value?	14.1.2				
Was a laboratory control sample analyzed with each batch at a minimum rate of 10% of samples?	14.1.4				